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Vinegar and Baking Soda #11. Pour 30 ml of vinegar into your mixing glass. 2. Take the temperature of the vinegar and record it on your lab sheet. 3. Look at the physical properties of the vinegar and baking soda and record your hypothesis on your lab sheet. 4. Mix 1 teaspoon of the solute (baking soda) into the mixing glass and stir. Set the timer for one minute, as it counts down, record your obse...

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[06 - DOUBLE REPLACEMENT](#)

Date: Double Replacement Reactions 1. Silver Nitrate is mixed with Aluminum Chloride. $1. 3\text{AgNO}_3(\text{aq}) + \text{AlCl}_3(\text{aq}) \rightarrow 3\text{AgCl}(\text{s}) + \text{Al}(\text{NO}_3)_3(\text{aq})$ 2. Potassium Iodide is mixed Zinc Acetate. $2. \text{KI}(\text{aq}) + \text{Zn}(\text{C}_2\text{H}_3\text{O}_2)_2 \rightarrow \text{NR}$ 3. Lead(II) Nitrate is mixed with Sodium Hydroxide. $3. \text{Pb}(\text{NO}_3)_2(\text{aq}) + 2\text{NaOH}(\text{aq}) \rightarrow \text{Pb}(\text{OH})_2(\text{s}) + 2\text{NaNO}_3(\text{aq})$ 4. Aluminum Chloride and Sodium Phosphate are mixed. $4. \text{AlCl}_3(\text{aq}) + \text{Na}_3\text{PO}_4(\text{aq}) \rightarrow \dots$

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